

REB, The Course

Class 6 Practice Assignment Calculations

Equation numbers correspond to Class 6 Practice Assignment Solution

Rate expressions for the mechanistic steps:

$$r_j = k_{j,f} \prod_{i_r} [i_r]^{-\nu_{i_r,j}} - k_{j,r} \prod_{i_p} [i_p]^{\nu_{i_p,j}}$$

$$r_2 = k_{2,f} [E] [S] - k_{2,r} [E-S] \quad (5)$$

$$r_3 = k_{3,f} [E-S] \quad (6)$$

$$r_4 = k_{4,f} [E-P] - k_{4,r} [E] [P] \quad (7)$$

Rate expression for the non-elementary reaction:

$$r_{i,j_{non}} = \sum_{j'} \nu_{i,j'} r_{j'}$$

$$r_1 = r_P = r_4 \quad (8)$$

Bodenstein steady-state approximations:

$$0 = \sum_{j'} \nu_{i_{RI},j'} r_{j'}$$

$$0 = -r_2 + r_4 \quad (9)$$

$$0 = r_2 - r_3 \quad (10)$$

$$0 = r_3 - r_4 \quad (11)$$

Conservation of enzyme:

$$C_{cat,0} = C_{cat,free} + \sum_{i_c} \kappa_{i_c} C_{i_c}$$

$$E_0 = [E] + [E-S] + [E-P] \quad (12)$$

Substitution of equations ((5) through (7) into equations (8), (10), and (11) yields equations (A) - (C).

$$r_P = k_{4,f} [E-P] - k_{4,r} [E] [P] \quad (\text{A})$$

$$0 = k_{2,f} [E] [S] - k_{2,r} [E-S] - k_{3,f} [E-S] \quad (\text{B})$$

$$0 = k_{3,f} [E-S] - k_{4,f} [E-P] + k_{4,r} [E] [P] \quad (\text{C})$$

Solving equation (B) yields equation (D).

$$[E-S] = \frac{k_{2,f} [S]}{k_{2,r} + k_{3,f}} [E] \quad (\text{D})$$

Solving equation (C) yields equation (E).

$$[E-P] = \frac{k_{3,f} [E-S] + k_{4,r} [E] [P]}{k_{4,f}} \quad (\text{E})$$

Substitution of equation (D) into equation (E) leads to equation (F).

$$[E-P] = \frac{k_{3,f} \frac{k_{2,f} [S]}{k_{2,r} + k_{3,f}} [E] + k_{4,r} [E] [P]}{k_{4,f}}$$

$$[E-P] = \left(\frac{\frac{k_{2,f} k_{3,f} [S]}{k_{2,r} + k_{3,f}} + k_{4,r} [P]}{k_{4,f}} \right) [E]$$

$$[E-P] = \left(\frac{k_{2,f} k_{3,f} [S]}{k_{4,f} (k_{2,r} + k_{3,f})} + \frac{k_{4,r}}{k_{4,f}} [P] \right) [E]$$

$$[E-P] = \frac{k_{2,f}k_{3,f}[S] + k_{4,r}(k_{2,r} + k_{3,f})[P]}{k_{4,f}(k_{2,r} + k_{3,f})} [E] \quad (F)$$

Substitution of equations (D) and (F) into equation (12) leads to an expression for the concentration of E.

$$\begin{aligned} E_0 &= [E] + \frac{k_{2,f}[S]}{k_{2,r} + k_{3,f}} [E] + \frac{k_{2,f}k_{3,f}[S] + k_{4,r}(k_{2,r} + k_{3,f})[P]}{k_{4,f}(k_{2,r} + k_{3,f})} [E] \\ E_0 &= \left(1 + \frac{k_{2,f}[S]}{k_{2,r} + k_{3,f}} + \frac{k_{2,f}k_{3,f}[S] + k_{4,r}(k_{2,r} + k_{3,f})[P]}{k_{4,f}(k_{2,r} + k_{3,f})} \right) [E] \\ E_0 &= \frac{k_{4,f}(k_{2,r} + k_{3,f}) + k_{2,f}k_{4,f}[S] + k_{2,f}k_{3,f}[S] + k_{4,r}(k_{2,r} + k_{3,f})[P]}{k_{4,f}(k_{2,r} + k_{3,f})} [E] \\ E &= \frac{k_{4,f}(k_{2,r} + k_{3,f}) E_0}{\left(k_{4,f}(k_{2,r} + k_{3,f}) + k_{2,f}(k_{3,f} + k_{4,f})[S] \right.} \\ &\quad \left. + k_{4,r}(k_{2,r} + k_{3,f})[P] \right) \end{aligned} \quad (13)$$

Substitution of equation (13) into equation (F) yields an expression for the concentration of E-P.

$$[E-P] = \frac{k_{2,f}k_{3,f}[S] + k_{4,r}(k_{2,r} + k_{3,f})[P]}{k_{4,f}(k_{2,r} + k_{3,f})} \frac{k_{4,f}(k_{2,r} + k_{3,f}) E_0}{\left(k_{4,f}(k_{2,r} + k_{3,f}) + k_{2,f}(k_{3,f} + k_{4,f})[S] \right.} \\ \left. + k_{4,r}(k_{2,r} + k_{3,f})[P] \right)$$

$$[E-P] = \frac{k_{2,f}k_{3,f}E_0 [S] + k_{4,r}(k_{2,r} + k_{3,f}) E_0 [P]}{\left(k_{4,f}(k_{2,r} + k_{3,f}) + k_{2,f}(k_{3,f} + k_{4,f}) [S] \right) + k_{4,r}(k_{2,r} + k_{3,f}) [P]} \quad (14)$$

Substitution of equations (13) and (14) in equation (A) leads to an acceptable form of the rate expression.

$$r_P = k_{4,f} \frac{k_{2,f}k_{3,f}E_0 [S] + k_{4,r}(k_{2,r} + k_{3,f}) E_0 [P]}{\left(k_{4,f}(k_{2,r} + k_{3,f}) + k_{2,f}(k_{3,f} + k_{4,f}) [S] \right) + k_{4,r}(k_{2,r} + k_{3,f}) [P]} - k_{4,r} \frac{k_{4,f}(k_{2,r} + k_{3,f}) E_0}{\left(k_{4,f}(k_{2,r} + k_{3,f}) + k_{2,f}(k_{3,f} + k_{4,f}) [S] \right) + k_{4,r}(k_{2,r} + k_{3,f}) [P]} [P]$$

$$r_P = \frac{k_{2,f}k_{3,f}k_{4,f}E_0 [S] + k_{4,r}k_{4,f}(k_{2,r} + k_{3,f}) E_0 [P] - k_{4,r}k_{4,f}(k_{2,r} + k_{3,f}) E_0 [P]}{\left(k_{4,f}(k_{2,r} + k_{3,f}) + k_{2,f}(k_{3,f} + k_{4,f}) [S] \right) + k_{4,r}(k_{2,r} + k_{3,f}) [P]}$$

$$r_P = \frac{k_{2,f}k_{3,f}k_{4,f}E_0 [S]}{\left(k_{4,f}(k_{2,r} + k_{3,f}) + k_{2,f}(k_{3,f} + k_{4,f}) [S] \right) + k_{4,r}(k_{2,r} + k_{3,f}) [P]} \quad (15)$$